

## **MATERIAL SAFETY DATA SHEET**

# 2330

Product Name DETECTAGAS 5PPM HYDROGEN CHLORIDE IN NITROGEN 58L

## 1. IDENTIFICATION OF THE MATERIAL AND SUPPLIER

Supplier Name BOC LIMITED (AUSTRALIA)

Address 10 Julius Avenue, North Ryde, NSW, AUSTRALIA, 2113

**Telephone** 131 262, (02) 8874 4400 **Fax** 132 427 (24 hours)

**Emergency** 1800 653 572 (24/7) (Australia only)

Web Site http://www.boc.com.au/

Synonym(s) 2330 - MSDS NUMBER • PRODUCT CODE: 29473NR • SPECIAL GAS MIXTURE

Use(s) CALIBRATION • INDUSTRIAL APPLICATIONS

**SDS Date** 11 Feb 2010

#### 2. HAZARDS IDENTIFICATION

NOT CLASSIFIED AS HAZARDOUS ACCORDING TO ASCC CRITERIA

CLASSIFIED AS A DANGEROUS GOOD BY THE CRITERIA OF THE ADG CODE

UN No. 1956 DG Class 2.2 Subsidiary Risk(s) None Allocated

Packing Group None Allocated Hazchem Code 2TE EPG 2C1

#### 3. COMPOSITION/ INFORMATION ON INGREDIENTS

Ingredient	Formula	CAS No.	Content v/v
HYDROGEN CHLORIDE	CI-H	7647-01-0	0.0005%
NITROGEN	N2	7727-37-9	remainder

## 4. FIRST AID MEASURES

Eye If in eyes, hold eyelids apart and flush the eye continuously with running water. Continue flushing until advised to

stop by the Poisons Information Centre or a doctor, or for at least 15 minutes.

Inhalation If inhaled, remove from contaminated area. To protect rescuer, use an Air-line respirator or Self Contained

Breathing Apparatus (SCBA). Apply artificial respiration if not breathing. Give oxygen if available. For advice,

contact a Poisons Information Centre on 13 11 26 (Australia Wide) or a doctor.

Skin If skin or hair contact occurs, remove contaminated clothing and flush skin and hair with running water. Continue

flushing with water until advised to stop by the Poisons Information Centre or a doctor.

**Ingestion** Due to product form and application, ingestion is considered unlikely.

Advice to Doctor Treat symptomatically

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**Product Name** 

**Explosion** 

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#### 5. FIRE FIGHTING MEASURES

Flammability Non flammable.

Fire and Temperatures in a fire may cause cylinders to rupture. Cool cylinders or containers exposed to fire by applying

water from a protected location. Do not approach cylinders or containers suspected of being hot.

**Extinguishing** Use water fog to cool containers from protected area.

Hazchem Code 2TE

#### 6. ACCIDENTAL RELEASE MEASURES

Spillage If the cylinder is leaking, evacuate area of personnel. Inform manufacturer/supplier of leak. Use personal protective equipment. Carefully move material to a well ventilated remote area, then allow to discharge. Do not

attempt to repair leaking valve or cylinder safety devices.

## 7. STORAGE AND HANDLING

Storage Do not store near incompatible materials. Cylinders should be stored below 45°C in a secure area, upright and

restrained to prevent cylinders from falling. Cylinders should also be stored in a dry, well ventilated area constructed of non-combustible material with firm level floor (preferably concrete), away from areas of heavy traffic

and emergency exits.

**Handling** Use of safe work practices are recommended to avoid eye or skin contact and inhalation. Do not drag, drop, slide

or roll cylinders. The uncontrolled release of a gas under pressure may cause physical harm. Use a suitable hand

truck for cylinder movement.

## 8. EXPOSURE CONTROLS/ PERSONAL PROTECTION

**Exposure Stds** 

Ingredient	Deference	TWA		STEL		
	Reference	ppm	mg/m3	ppm	mg/m3	
Hydrogen chloride	ASCC (AUS)	5	7.5			
Nitrogen	ASCC (AUS)	Asphyxiant				

Biological Limits No biological limit allocated.

**Engineering Controls** 

**PPE** 

Avoid inhalation. Use in well ventilated areas. Where an inhalation risk exists, mechanical extraction ventilation is

recommended. Maintain vapour levels below the recommended exposure standard.

Wear safety or safety boots, cotton or leather gloves and safety glasses. In poorly ventilated areas or where a significant inhalation risk exists (ie. oxygen deficient atmosphere), wear a Full-face Air-line respirator or Self Contained Breathing Apparatus (SCBA).







## 9. PHYSICAL AND CHEMICAL PROPERTIES

**COLOURLESS GAS** NOT SOLUBLE **Appearance** Solubility (Water) **ODOURLESS Specific Gravity** NOT APPLICABLE Odour NOT APPLICABLE % Volatiles 100 % Vapour Pressure **NOT AVAILABLE** Flammability NON FLAMMABLE **Relative Density** Flash Point **NOT APPLICABLE** 1 (Air = 1)**Boiling Point NOT RELEVANT Upper Explosion Limit** NOT RELEVANT **Melting Point NOT RELEVANT Lower Explosion Limit NOT RELEVANT Evaporation Rate NOT APPLICABLE** 

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#### 10. STABILITY AND REACTIVITY

**Chemical Stability** Stable under recommended conditions of storage.

**Conditions to Avoid** Avoid shock, friction, heavy impact, heat, sparks, open flames and other ignition sources.

Material to AvoidIncompatible with alkalis (eg. hydroxides). May be corrosive to most metals.DecompositionNo hazardous products are formed by the decomposition of this material.

Hazardous Reactions Polymerization will not occur.

#### 11. TOXICOLOGICAL INFORMATION

#### Health Hazard Summary

Asphyxiant gas - irritant. May replace oxygen in the inhaled air and cause asphyxiation. As the amount of oxygen inhaled is reduced from 21-14% volume, the pulse rate may accelerate and the rate and volume of breathing may increase. The ability to maintain attention and think clearly is diminished, muscular co-ordination is somewhat disturbed. As oxygen decreases from 14-10% volume, judgement becomes faulty, severe injuries may result in no pain. Muscular effort may lead to rapid fatigue. Further reduction to 6% may result in nausea and vomiting. Ability to move may be lost. Permanent brain damage may result even after resuscitation from exposure to this low level of oxygen. Below 6% breathing is in gasps and convulsions may occur. Inhalation of a mixture containing no oxygen may result in unconsciousness from the first breath and death may follow in minutes. Hydrogen chloride causes eye and respiratory irritation. High level exposure may result in incoordination, vomiting, mental instability, lung damage, convulsions, coma and death.

Eye Irritant. Contact may result in irritation, lacrimation, pain and redness.

Inhalation Irritant gas - asphyxiant. Effects are proportional to oxygen displacement with symptoms of air hunger, rapid

breathing, elevated heart rate, drowsiness and loss of mental alertness. Hydrogen chloride causes eye and respiratory irritation. High level exposure may result in incoordination, vomiting, mental instability, lung damage,

convulsions, coma and death.

**Skin** Irritant. Contact may result in irritation.

**Ingestion** Ingestion is considered unlikely due to product form. Inhalation may lead to irritation of mouth and throat.

Toxicity Data HYDROGEN CHLORIDE (7647-01-0) LC50 (Inhalation): 3124 ppm/1 hour (rat)

## 12. ECOLOGICAL INFORMATION

**Environment** 

Limited ecotoxicity data was available for this product at the time this report was prepared. Ensure appropriate measures are taken to prevent this product from entering the environment.

#### 13. DISPOSAL CONSIDERATIONS

Waste Disposal Cylinders should be returned to the manufacturer or supplier for disposal of contents.

**Legislation** Dispose of in accordance with relevant local legislation.

#### 14. TRANSPORT INFORMATION

#### **Transport**

Ensure cylinder is separated from driver and that outlet of relief device is not obstructed. Refer to Commonwealth, State and Territory Dangerous Goods Legislation which contain requirements which affect gas storage and transport.



#### CLASSIFIED AS A DANGEROUS GOOD BY THE CRITERIA OF THE ADG CODE

Shipping Name COMPRESSED GAS, N.O.S. (Contains Nitrogen)

UN No. 1956 DG Class 2.2 Subsidiary Risk(s) None Allocated

Packing Group None Allocated Hazchem Code 2TE EPG 2C1

## 15. REGULATORY INFORMATION

Poison Schedule A poison schedule number has not been allocated to this product using the criteria in the Standard for the Uniform

Scheduling of Drugs and Poisons (SUSDP).

All chemicals listed on the Australian Inventory of Chemical Substances (AICS).

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CHEM ALERT

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### 16. OTHER INFORMATION

# Additional Information

The storage of significant quantities of gas cylinders must comply with AS4332 The storage and handling of gases in cylinders.

ASPHYXIANT GASES: Asphyxiant gases may displace oxygen, leading to oxygen deficiency. Where oxygen content is low, effects may include: 12-16% oxygen: increased breathing/ pulse rate, lack of coordination; 10-14%: mental disturbance, fatigue, breathing stress; 6-10%: vomiting, collapse and possible unconsciousness; 0-6%: convulsions, respiratory collapse and death.

ASPHYXIANTS (1): When present in the atmospheres in high concentrations, asphyxiants reduce the oxygen concentration by displacement. Atmospheres deficient in oxygen do not provide adequate sensory warning of danger and most simple asphyxiants are odourless. Therefore it is not appropriate to recommend an exposure standard for each asphyxiant, but to maintain oxygen concentrations. However, some asphyxiants may be given an exposure standard due to the potential for narcotic effects at high concentrations or an explosion hazard.

ASPHYXIANTS (2): There is a significant hazard associated with workers entering poorly ventilated areas (eg. tanks) where oxygen may be deficient. An air supplied breathing apparatus may be required if adequate ventilation is not ensured.

#### ABBREVIATIONS:

ADB - Air-Dry Basis.

BEI - Biological Exposure Indice(s)

CAS# - Chemical Abstract Service number - used to uniquely identify chemical compounds.

CNS - Central Nervous System.

EINECS - European INventory of Existing Commercial chemical Substances.

IARC - International Agency for Research on Cancer.

M - moles per litre, a unit of concentration.

mg/m3 - Milligrams per cubic metre.

NOS - Not Otherwise Specified.

NTP - National Toxicology Program.

OSHA - Occupational Safety and Health Administration.

pH - relates to hydrogen ion concentration using a scale of 0 (high acidic) to 14 (highly alkaline).

ppm - Parts Per Million.

RTECS - Registry of Toxic Effects of Chemical Substances.

TWA/ES - Time Weighted Average or Exposure Standard.

#### **HEALTH EFFECTS FROM EXPOSURE:**

It should be noted that the effects from exposure to this product will depend on several factors including: frequency and duration of use; quantity used; effectiveness of control measures; protective equipment used and method of application. Given that it is impractical to prepare a Chem Alert report which would encompass all possible scenarios, it is anticipated that users will assess the risks and apply control methods where appropriate.

#### PERSONAL PROTECTIVE EQUIPMENT GUIDELINES:

The recommendation for protective equipment contained within this Chem Alert report is provided as a guide only. Factors such as method of application, working environment, quantity used, product concentration and the availability of engineering controls should be considered before final selection of personal protective equipment is made.

## **Report Status**

This document has been compiled by RMT on behalf of the manufacturer of the product and serves as the manufacturer's Safety Data Sheet ('SDS').

It is based on information concerning the product which has been provided to RMT by the manufacturer or obtained from third party sources and is believed to represent the current state of knowledge as to the appropriate safety and handling precautions for the product at the time of issue. Further clarification regarding any aspect of the product should be obtained directly from the manufacturer.

While RMT has taken all due care to include accurate and up-to-date information in this SDS, it does not provide any warranty as to accuracy or completeness. As far as lawfully possible, RMT accepts no liability for any loss, injury or damage (including consequential loss) which may be suffered or incurred by any person as a consequence of their reliance on the information contained in this SDS.

#### **Prepared By**

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#### **Product Name**

## **DETECTAGAS 5PPM HYDROGEN CHLORIDE IN NITROGEN 58L**

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