

SAFETY DATA SHEET

1242

Product Name 4 COMPONENT MIXTURE (CHF3, CO2, N2O, BALANCE O2)

1. IDENTIFICATION OF THE MATERIAL AND SUPPLIER

Supplier Name BOC LIMITED (AUSTRALIA)

Address 10 Julius Avenue, North Ryde, NSW, AUSTRALIA, 2113

Telephone 131 262, (02) 8874 4400 **Fax** 132 427 (24 hours)

Emergency 1800 653 572 (24/7) (Australia only)

Web Site http://www.boc.com.au/

Synonym(s) 1242 - SDS NUMBER • PRODUCT CODE: 285, 288 • SPECIAL GAS MIXTURE

Use(s) CALIBRATION • INDUSTRIAL APPLICATIONS

SDS Date 19 Apr 2010

2. HAZARDS IDENTIFICATION

CLASSIFIED AS HAZARDOUS ACCORDING TO ASCC CRITERIA

RISK PHRASES

R61 May cause harm to the unborn child.

SAFETY PHRASES

S51 Use only in well ventilated areas.

CLASSIFIED AS A DANGEROUS GOOD BY THE CRITERIA OF THE ADG CODE

UN No. 3156 DG Class 2.2 Subsidiary Risk(s) 5.1

Packing Group None Allocated Hazchem Code 2S EPG 2B1

3. COMPOSITION/ INFORMATION ON INGREDIENTS

Ingredient	Formula	CAS No.	Content
NITROUS OXIDE	N20	10024-97-2	<50%
OXYGEN	O2	7782-44-7	remainder
CARBON DIOXIDE	CO2	124-38-9	<6%
TRIFLUOROMETHANE	C-H-F3	75-46-7	<5%

4. FIRST AID MEASURES

Eye None required.

Inhalation If inhaled, remove from contaminated area. Apply artificial respiration if not breathing. For advice, contact a

Poisons Information Centre on 13 11 26 (Australia Wide) or a doctor.

Skin None required.

Ingestion Due to product form and application, ingestion is considered unlikely.

Advice to Doctor Treatment for hyperoxia.



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5. FIRE FIGHTING MEASURES

Flammability Non flammable - oxidising agent. Supports combustion and may cause fire/explosion in contact with incompatible

substances, strong acids, reducing agents, combustibles and flammables.

Fire and **Explosion**

Extinguishing

Temperatures in a fire may cause cylinders to rupture and internal pressure relief devices to be activated. Cool cylinders or containers exposed to fire by applying water from a protected location. Do not approach cylinders or containers suspected of being hot. Remove cool cylinders from the path of the fire if safe to do so. Ensure working area is well ventilated before re-use Notify the nearest BOC centre that you will be returning a faulty cylinder. Residual product will be disposed of when the cylinder is returned.

Use water fog to cool containers from protected area.

Hazchem Code

6. ACCIDENTAL RELEASE MEASURES

If the cylinder is leaking, evacuate area of personnel. Inform manufacturer/supplier of leak. Use personal Spillage protective equipment. Carefully move material to a well ventilated remote area, then allow to discharge. Do not

attempt to repair leaking valve or cylinder safety devices.

7. STORAGE AND HANDLING

Do not store near sources of ignition or incompatible materials. Cylinders should be stored below 45°C in a secure Storage area, upright and restrained to prevent cylinders from falling. Cylinders should also be stored in a dry, well ventilated area constructed of non-combustible material with firm level floor (preferably concrete), away from areas

of heavy traffic and emergency exits.

Handling Use of safe work practices are recommended to avoid inhalation. Do not drag, drop, slide or roll cylinders. The

uncontrolled release of a gas under pressure may cause physical harm. Use a suitable hand truck for cylinder

movement.

8. EXPOSURE CONTROLS/ PERSONAL PROTECTION

Exposure Stds

Ingredient	Poforono	TWA		STEL	
	Reference	ppm	mg/m3	ppm	mg/m3
Carbon dioxide	ASCC (AUS)	5000	9000	30000	54000
Carbon dioxide in coal mines	ASCC (AUS)	12500	22500	30000	54000
Nitrous oxide	ASCC (AUS)	25	45		

Biological Limits No biological limit allocated.

Engineering Controls

Avoid inhalation. Use in well ventilated areas. Where an inhalation risk exists, mechanical extraction ventilation is recommended. Maintain vapour levels below the recommended exposure standard.

PPE Wear safety boots, leather gloves and safety glasses. Where an inhalation risk exists, wear: an Air-line respirator or self Contained Breathing Apparatus (SCBA).





NOT APPLICABLE



9. PHYSICAL AND CHEMICAL PROPERTIES

COLOURLESS GAS Solubility (Water) 0.032 L/L (Oxygen) **Appearance** Odour SLIGHT SWEET ODOUR **Specific Gravity** NOT APPLICABLE Ηα **NOT APPLICABLE** % Volatiles 100 %

Vapour Pressure **NOT AVAILABLE Flammability** NON FLAMMABLE **Vapour Density** 1.3 (Air = 1) (Approximately) **Flash Point** NOT RELEVANT **Boiling Point NOT AVAILABLE** NOT RELEVANT **Upper Explosion Limit Melting Point NOT AVAILABLE Lower Explosion Limit NOT RELEVANT**

Cylinder Pressure 7000 - 12000 kPa @ 15°C



Evaporation Rate

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10. STABILITY AND REACTIVITY

Material to Avoid Combustible materials such as oil and grease can spontaneously ignite at low temperatures in oxygen

enriched atmospheres. Materials which burn in air, will burn more vigorously in oxygen enriched atmospheres. Nitrous oxide will react with powerful reducing agents such as phosphine, stannous chloride and hydrogen. Ensure that all materials and lubricants in contact with this product are oxygen compatible.

All non-metals must be oxygen compatible. Copper is the most commonly used metal.

Decomposition May evolve toxic gases if heated to decomposition.

Hazardous Reactions Polymerization will not occur.

11. TOXICOLOGICAL INFORMATION

Health Hazard Summary Anaesthetic - non irritating gas. Nitrous oxide passes into all gas containing spaces in the body faster than nitrogen passes out, thus it should not be used with any condition where its expansion might be dangerous. May induce vomiting in susceptible individuals. Epidemiological studies suggest an increased risk of spontaneous abortion and low birth weight in off-spring in female workers employed in operating theatres and dental surgeries.

These findings are controversial.

Eye Non irritant

Inhalation Non irritant - anaesthetic. After prolonged use nitrous oxide may inactivate vitamin B12, resulting in megaloblastic

changes in bone marrow. Exceptionally heavy occupational exposure and addiction may result in myeloneuropathy and sub-acute combined degeneration. This would require levels in excess of 5,000ppm to

frequent (more than once every two days) exposure to analgesic concentrations.

Skin Non irritant.

Ingestion Ingestion is considered unlikely due to product form.

Toxicity Data NITROUS OXIDE (10024-97-2)

LC50 (Inhalation): 1068 mg/m3 (rat)

TCLo (Inhalation): 1 pph/8 hours (rat - reproductive effects)

CARBON DIOXIDE (124-38-9)

LC50 (Inhalation): 470000 ppm/30M (rat) LCLo (Inhalation): 9 pph/5M (human)

12. ECOLOGICAL INFORMATION

Environment

Limited ecotoxicity data was available for this product at the time this report was prepared. Ensure appropriate measures are taken to prevent this product from entering the environment.

13. DISPOSAL CONSIDERATIONS

Waste Disposal Cylinders should be returned to the manufacturer or supplier for disposal of contents.

Legislation Dispose of in accordance with relevant local legislation.

14. TRANSPORT INFORMATION

Transport Ensure cylinder is separated from driver and that outlet of relief device is not obstructed.





CLASSIFIED AS A DANGEROUS GOOD BY THE CRITERIA OF THE ADG CODE

Shipping Name COMPRESSED GAS, OXIDIZING, N.O.S.

UN No. 3156 DG Class 2.2 Subsidiary Risk(s) 5.1

Packing Group None Allocated Hazchem Code 2S EPG 2B1

15. REGULATORY INFORMATION

Poison Schedule Classified as a Schedule 4 (S4) Poison using the criteria in the Standard for the Uniform Scheduling of Drugs and

Poisons (SUSDP).

AICS All chemicals listed on the Australian Inventory of Chemical Substances (AICS).

ChemAlert.

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16. OTHER INFORMATION

Additional Information

The storage of significant quantities of gas cylinders must comply with AS4332 The storage and handling of gases in cylinders.

Application Method: Gas regulator of suitable pressure and flow rating fitted to cylinder valve or manifold with low pressure gas distribution to equipment.

ABBREVIATIONS:

ADB - Air-Dry Basis.

BEI - Biological Exposure Indice(s)

CAS# - Chemical Abstract Service number - used to uniquely identify chemical compounds.

CNS - Central Nervous System.

EINECS - European INventory of Existing Commercial chemical Substances.

IARC - International Agency for Research on Cancer.

M - moles per litre, a unit of concentration.

mg/m3 - Milligrams per cubic metre.

NOS - Not Otherwise Specified.

NTP - National Toxicology Program.

OSHA - Occupational Safety and Health Administration.

pH - relates to hydrogen ion concentration using a scale of 0 (high acidic) to 14 (highly alkaline).

ppm - Parts Per Million.

RTECS - Registry of Toxic Effects of Chemical Substances.

TWA/ES - Time Weighted Average or Exposure Standard.

HEALTH EFFECTS FROM EXPOSURE:

It should be noted that the effects from exposure to this product will depend on several factors including: frequency and duration of use; quantity used; effectiveness of control measures; protective equipment used and method of application. Given that it is impractical to prepare a Chem Alert report which would encompass all possible scenarios, it is anticipated that users will assess the risks and apply control methods where appropriate.

PERSONAL PROTECTIVE EQUIPMENT GUIDELINES:

The recommendation for protective equipment contained within this Chem Alert report is provided as a guide only. Factors such as method of application, working environment, quantity used, product concentration and the availability of engineering controls should be considered before final selection of personal protective equipment is made.

Report Status

This document has been compiled by RMT on behalf of the manufacturer of the product and serves as the manufacturer's Safety Data Sheet ('SDS').

It is based on information concerning the product which has been provided to RMT by the manufacturer or obtained from third party sources and is believed to represent the current state of knowledge as to the appropriate safety and handling precautions for the product at the time of issue. Further clarification regarding any aspect of the product should be obtained directly from the manufacturer.

While RMT has taken all due care to include accurate and up-to-date information in this SDS, it does not provide any warranty as to accuracy or completeness. As far as lawfully possible, RMT accepts no liability for any loss, injury or damage (including consequential loss) which may be suffered or incurred by any person as a consequence of their reliance on the information contained in this SDS.

Prepared By

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